

Species Tag:	53006	Name:	Cl-37-O-v1
Version:	1		Chlorine monoxide,
Date:	Jan. 1991		X $^2\Pi$ states, v = 1
Contributor:	E. A. Cohen		
Lines Listed:	2132	Q(300.0)=	3409.573
Freq. (GHz) <	2834	Q(225.0)=	2347.470
Max. J:	81	Q(150.0)=	1449.105
LOGSTR0=	-9.9	Q(75.00)=	700.971
LOGSTR1=	-11.5	Q(37.50)=	357.108
Isotope Corr.:	-0.611	Q(18.75)=	186.209
Egy. ( $\text{cm}^{-1}$ ) >	837.1	Q(9.375)=	101.135
$\mu_a$ =	1.2758	A=	
$\mu_b$ =		B=	18113.451
$\mu_c$ =		C=	

The experimental lines are from R. K. Kakar, E. A. Cohen, and M. Geller, 1978, J. Mol. Spect. **70**, 243, and E. A. Cohen, H. M. Pickett, and M. Geller, 1984, J. Mol. Spect. **106**, 430. The millimeter and submillimeter data were combined with the FT infrared data of J. B. Burkholder *et al.*, 1987, J. Mol. Spect. **124**, 139, in a simultaneous fit to both vibrational states. The partition function is a sum over the ground and first vibrational state up to F = 90 for both the  $\Omega = 3/2$  and  $\Omega = 1/2$  states. The dipole moment is from D. Yaron, K. Peterson, and W. Klemperer, 1988, J. Chem. Phys. **88**, 4702, and assumed to be unchanged for  $^{37}\text{ClO}$ .